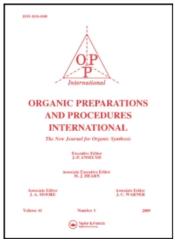
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SYNTHESIS OF 3-MERCAPTOTHIOPHENE-2-AND 4-CARBOXYLIC ACIDS

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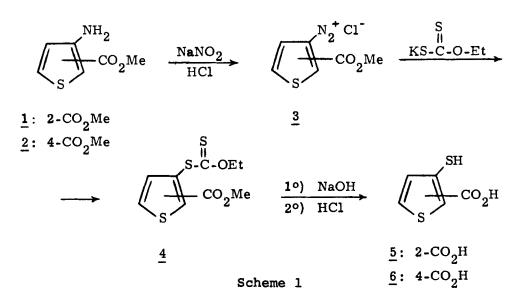
SYNTHESIS OF 3-MERCAPTOTHIOPHENE-2-AND 4-CARBOXYLIC ACIDS

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A recent paper ¹ described the synthesis of 3-mercaptothiophene-2-carboxylic acid ($\underline{5}$) by the Newman Karnes phenol-thiophenol conversion applied to 3-hydroxy-2-methoxycarbonylthiophene. However, 3-mercaptothiophene-4-carboxylic acid ($\underline{6}$), could not be synthesized by the same procedure, because the starting material, 3-hydroxy-4-methoxycarbonylthiophene is extremely labile to bases.²

We report here the first synthesis of compound <u>6</u> and an improved preparation of compound <u>5</u> from the corresponding 3-amino-2 or 4methoxycarbonylthiophenes (1, 2). The method (Scheme 1) is based on the

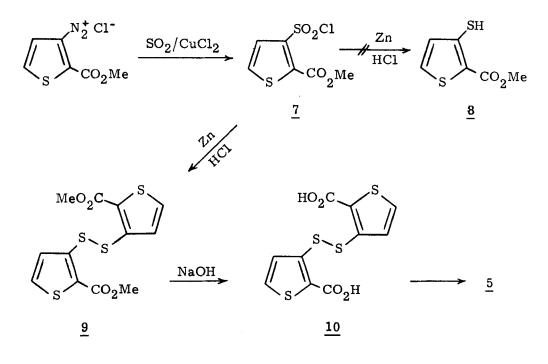


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facile conversion of the 3-aminothiophene derivatives $\underline{1}$ and $\underline{2}$ into stable diazonium salts $\underline{3}$ which upon treatment with potassium ethyl xanthate gave $\underline{4}$. These compounds were not isolated but were hydrolyzed with alkali, to 5 (96% yield) and 6 (65% yield) respectively.

Another method attempted (Scheme 2) for the synthesis of 5 gave poorer yields of this compound. The diazonium salt of 3-amino-2-methoxycarbonylthiophene was converted to the chlorosulfonyl derivative 7 by the Meerwein reaction. Compound 7 however, upon reduction gave the



Scheme 2

disulfide <u>9</u> instead of the mercaptan <u>8</u>. The structural assignment for <u>9</u> was based on the IR (absence of SH group) and mass spectrum which showed a molecular ion at 346 m/e. This behaviour parallels that observed by Paulmier³ in the synthesis of 3-mercapto-2-acetylthiophene. Compound <u>9</u> could be hydrolyzed to the diacid <u>10</u> which was reduced to <u>5</u> with zinc and hydrochloric acid, in a low overall yield.

EXPERIMENTAL SECTION

Melting points were determined on a Büchi 510 melting point apparatus and are uncorrected. Mass spectra were recorded on a Hitachi Perkin-Elmer RMU-GMG mass spectrometer. IR spectra were recorded on a Perkin-Elmer 257 spectrometer. H-NMR spectra were recorded on a Varian EM 390 90 MHz spectrometer, with TMS as internal standard.

<u>3-Mercaptothiophene-2-carboxylic acid</u> (5).- 3-Amino-2-methoxycarbonylthiophene⁴ (2.5 g, 16 mmol) was added gradually to a vigorously stirred 6M hydrochloric acid solution (12 ml). The reaction mixture was stirred at room temperature for 30 min. It was then cooled below 0° (ice-salt bath), it was diazotised with NaNO₂ (1.1 g, 16 mmol) in H₂O (8 ml) and the resulting diazonium salt was stirred another hour at this temperature and then added slowly onto a well-stirred solution of Na₂CO₃ (2.2 g, 19 mmol) and potassium ethyl xanthate (2.55 g, 16 mmol) in H₂O (25 ml) and heated at 60°. The temperature of the mixture was maintained at 60-70° until the evolution of nitrogen has ceased (about 1 hr.). After cooling, the mixture was extracted with ether (4x 30 ml). The ethereal extract was washed with 10% NaOH solution (20 ml) and H₂O (2x 20 ml), dried over MgSO₄ and evaporated to dryness.

To a solution of the resulting reddish oil in EtOH (15 ml) was added a solution of 4H NaOH (125 ml) and the mixture was refluxed for 4 hrs. After cooling, the solvent was removed, H_2O (25 ml) was added and this solution was extracted with ether (2x10 ml). The aqueous layer was acidified with 15% HCl to yield 2.4 g (96%) of a colourless solid, mp. 152-153° (dec.) (water).

Mass spectrum: $m/e = 160 (M^+)$. IR (nujol): 3200-2500 (OH); 2550 (SH); 1650 cm⁻¹ (C=O). ¹H-NMR (DMSO-d₆): δ 1.15 (s, 1H, SH); 6.27 (s, 1H, COOH); 7.20 (d, 1H, J \simeq 5 Hz, thiophene 4-H); 7.95 ppm (d, 1H, J \simeq 5 Hz, thiophene 5-H).

<u>Anal.</u> Calcd. for $C_5H_4O_2S_2$: C, 37.50; H, 2.50; S, 40.00 Found: C, 37.71; H, 2.43; S, 39.87 <u>3-Mercaptothiophene-4-carboxylic acid</u> (6). This compound, mp. 215-216° (dec.) (acetic acid), was obtained in a similar way as described for compound 5 starting from 3-amino-4-methoxycarbonylthiophene, ⁵ in a 65% yield.

Mass spectrum: $m/e = 160 (M^+)$. IR (nujol): 3300-2500 (OH); 2600 (SH); 1675 cm⁻¹ (C=O). ¹H-NMR (DMSO-d₆): δ 2.10 (s, 1H, SH); 3.90 (s, 1H, COOH); 7.40 (d, 1H, J \simeq 3Hz, thiophene 5-H); 8.40 ppm (d, 1H, J \simeq 3Hz, thiophene 2-H).

<u>Anal.</u> Calcd. for $C_5H_4O_2S_2$: C, 37.50; H, 2.50; S, 40.00 Found: C, 37.43; H, 2.48; S, 40.21

<u>3-Chlorosulfonyl-2-methoxycarbonylthiophene</u> (7). - The diazonium salt of 3-amino-2-methoxycarbonylthiophene (16 mmol) obtained as described previously was added to a freshly prepared solution obtained by addition of CuCl₂ (0.95 g) in H₂O (0.95 ml) to SO₂ (34.19 g) in AcOH. The resulting reaction mixture was poured onto ice-water (65-80 ml) and the solid so formed was collected washed with H₂O, dried and crystallized from n-hexane to yield 1.58 g (41%) of a colourless solid, mp. 63-64°.

IR (nujol): 1730 (C=O); 1375, 1200 cm⁻¹ (SO₂). ¹H-NMR (DMSOd₆): δ 4.10 (s, 3H, CH₃); 7.35 (d, 1H, J \simeq 5 Hz, thiophene 4-H); 7.75 (d, 1H, J \simeq 5 Hz, thiophene-5H).

<u>Anal.</u> Calcd. for C₆H₅ClO₄S₂: C, 29.94; H, 2.09; S, 13.32 Found: C, 29.83; H, 2.03; S, 12.89

Bis-[2-methoxycarbonyl-3-thienyl]-disulfide (9). - Compound 7 (4.0 g, 17 mmol) was added to a vigorously stirred solution of concentrated HCl (5.1 ml) in ice. Zinc dust (5.2 g, 80 mmol) was then added in portions; the temperature was maintained between 10-15°. The reaction mixture was stirred at room temperature for 4 hrs. and then extracted with ether (2x20 ml) to yield 3.23 g (56%) of a colourless solid, mp. 169-171° (ethanol), by evaporation of the ethereal extracts.

Mass spectrum: $m/e = 346 (M^+)$. IR (nujol): 1700 cm⁻¹ (C=O). ¹H-NMR (DMSO-d₆): δ 3.95 (s, 3H, CH₃); 7.40 (d, 1H, J \simeq 5 Hz, thiophene

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4-H); 7.55 (d, 1H, $J \simeq 5 Hz$, thiophene 5-H). <u>Anal</u>. Calcd. for $C_{12}H_{10}O_4S_4$: C, 41.60; H, 2.91; S, 37.01 Found: C, 41.62; H, 2.89; S, 37.04

<u>Bis-[2-carboxy-3-thienyl]-disulfide</u> (10). - A suspension of compound <u>9</u> (0.56 g, 1.6 mmol) in 2N NaOH (4.85 ml) was heated at reflux for 1 hr. After cooling, the reaction mixture was acidified (to pH 1) with 15% HCl to yield 0.46 g (90%) of a colourless solid, mp. 240-241° (ether-n-hexane).

Mass spectrum: m/e = 318 (M⁺). IR (nujol): 3100-2500 (OH); 1655 cm⁻¹ (C=O). ¹H-NMR (DMSO-d₆): 3.60 (s, 1H, COOH), 7.20 (d, 1H, J \simeq 5Hz, thiophene 4-H); 7.80 (d, 1H, J \simeq 5Hz, thiophene 5-H). <u>Anal.</u> Calcd. for C₁₀H₆O₄S₄: C, 37.72; H, 1.90; S, 40.28 Found: C, 37.47; H, 2.15; S, 39.99

<u>3-Mercaptothiophene-2-carboxylic acid</u> (5) from Compound <u>10</u>. – Zinc dust (1.8 g, 28 mmol) was added gradually to a well-stirred mixture of diacid <u>10</u> (1.5 g, 4.7 mmol), concentrated HCl (1.6 ml) and H_2O (1.6 ml). The reaction mixture was refluxed for 4 hrs., the hot solution was filtered rapidly and compound 5 (0.9 g, 60%) crystallized from the filtrate.

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